

Influence de la nature de la paroi sur les réactions de recombinaison des radicaux

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OUTLINE

- ❖ Introduction
- ❖ Kinetics of atoms in low pressure molecular plasmas
- ❖ How to determine the atom recombination frequency (γ) ?
- ❖ Example : determination of γ_O by time resolved optical emission spectroscopy (TROES)
 - influence of the nature of the wall on γ_O in pulsed oxygen ICP plasmas
 - Values of γ_O in O₂/TTIP and O₂/HMDSO plasmas
- ❖ Conclusion

Kinetics of X atoms in low pressure plasmas

$X_2 = Cl_2, O_2, H_2 \dots$

$X_2 + e \rightarrow X + X + e \quad K_{diss}$ (function of T_e)

$X + paroi \rightarrow \frac{1}{2} X_2 \quad K_{paroi} = \text{loss frequency of } X \text{ at the wall,}$

$K_{paroi} = 1/\tau, \quad \tau : \text{characteristic loss time}$

Global Model :

$$[X] = 2 K_{diss} \times n_e \times [X_2] / K_{paroi} = \tau \times K_{diss} \times n_e \times [X_2]$$

X_2 dissociation degree :

$$= \frac{[X]}{2[X_2] + [X]} = \frac{100}{1 + \frac{2K_{paroi}}{K_{diss} n_e}}$$

Recombination probability : γ

$$\tau = \tau_{diff} + \tau_{loss}$$

$$\tau = \frac{\Lambda^2}{D} + \frac{V}{A} \cdot \frac{2(2-\gamma)}{\bar{v} \cdot \gamma}$$

$$\frac{1}{\Lambda^2} = \left(\frac{\pi}{L}\right)^2 + \left(\frac{2.405}{R}\right)^2$$

D = binary diffusion coefficient of X in X₂

V = Volume of the reactor

A = Area of the reactor

V = thermal velocity of X

Chantry et al J. Appl. Phys. 62, 1141 (1987)

Dissociation degree of Cl₂ as a function of γ_{Cl}

V.M. Donnelly, J. Guha, L. Stafford, JVSTA. 010801(2011)

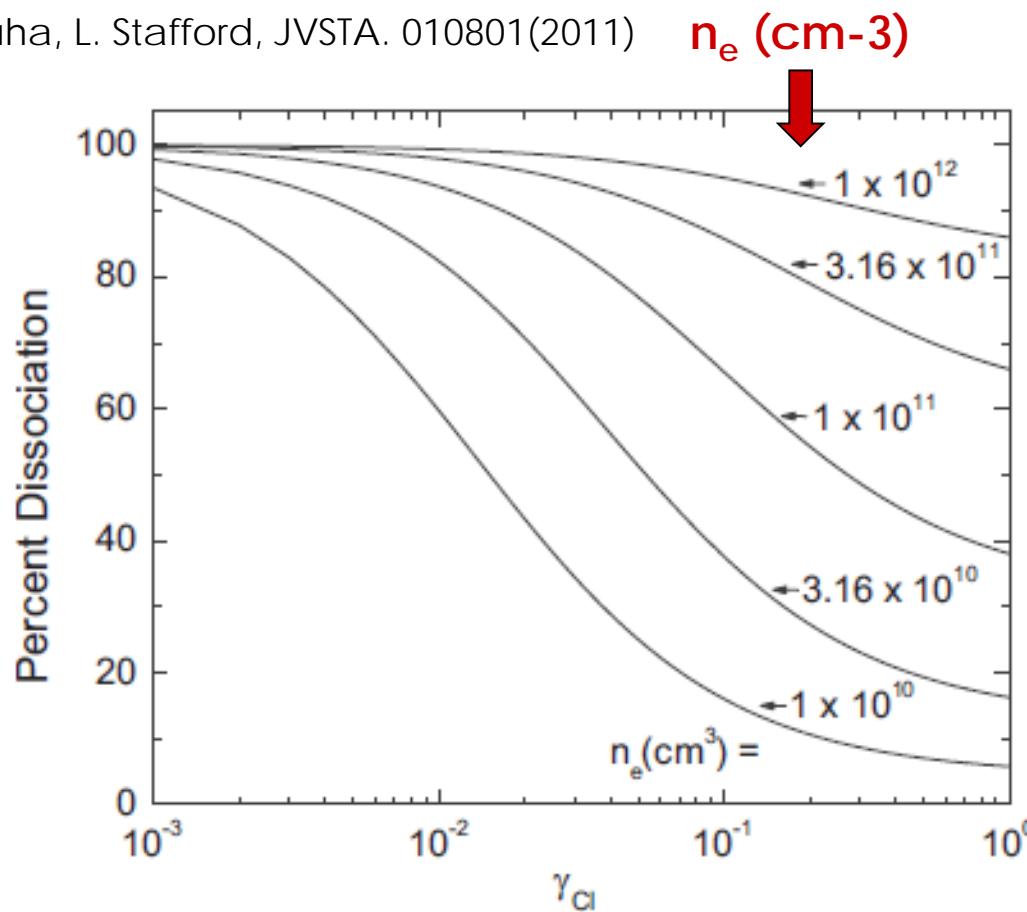
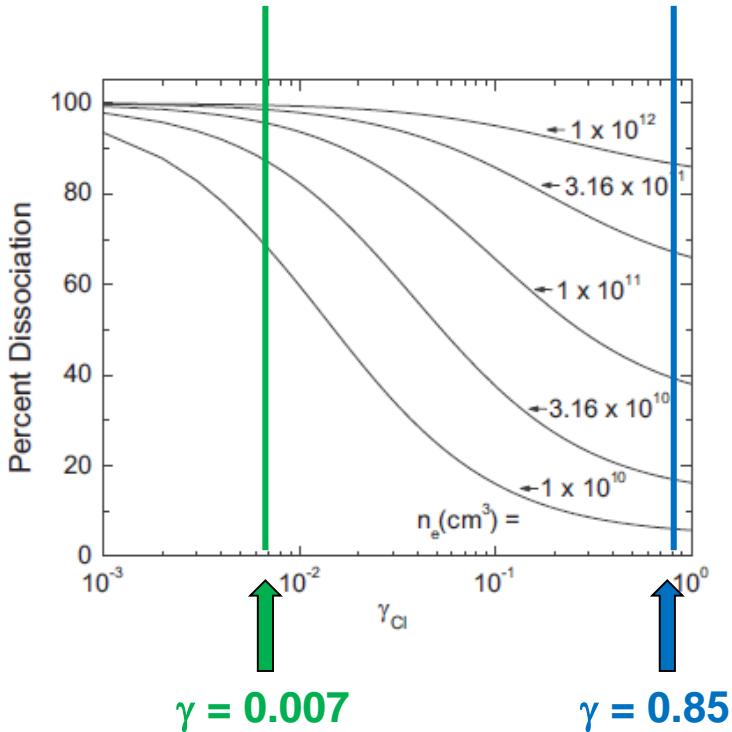


FIG. 1. Global model predictions of percent dissociation of Cl₂ in a chlorine plasma as a function of the Cl recombination coefficient for different electron densities. $R=L=20$ cm, pressure=10 mTorr, gas temperature, T_g =300 K, and T_e =3 eV.

Values of gamma for Cl on stainless steel (literature)



V.M. Donnelly, J. Guha, L. Stafford, JVSTA. 010801(2011)

Kota et al : value derived from a high vacuum beam experiment : $\gamma = 0.85$

Richards and Sawin : value derived from plasma measurements : $\gamma = 0.007$

γ dependent of :

- the nature of the material of the walls of the reactor
- the surface roughness (the specific area)
- the surface temperature
- the reactions at the wall : vacant sites created by ion bombardment.....

How to determine γ ?

best fit between predictions of the global model and experimental measurements in Cl₂ ICP plasmas

⇒ $\gamma = 0.04$ and 0.03 in two different studies (Malyshev et al, Ullal et al)

What happens at the walls ?

stainless steel walls are rapidly covered with silicon oxide or oxichloride (SiOCl) due to etching of the dielectric quartz windows of the ICP
⇒ surface quite different from stainless steel

General case :

γ depends on the nature of the wall surface, which itself depends on the plasma process and species present in the plasma

⇒ need to measure γ in situ, in « real » deposition conditions

Some examples of diagnostics used to determine in situ the value of γ

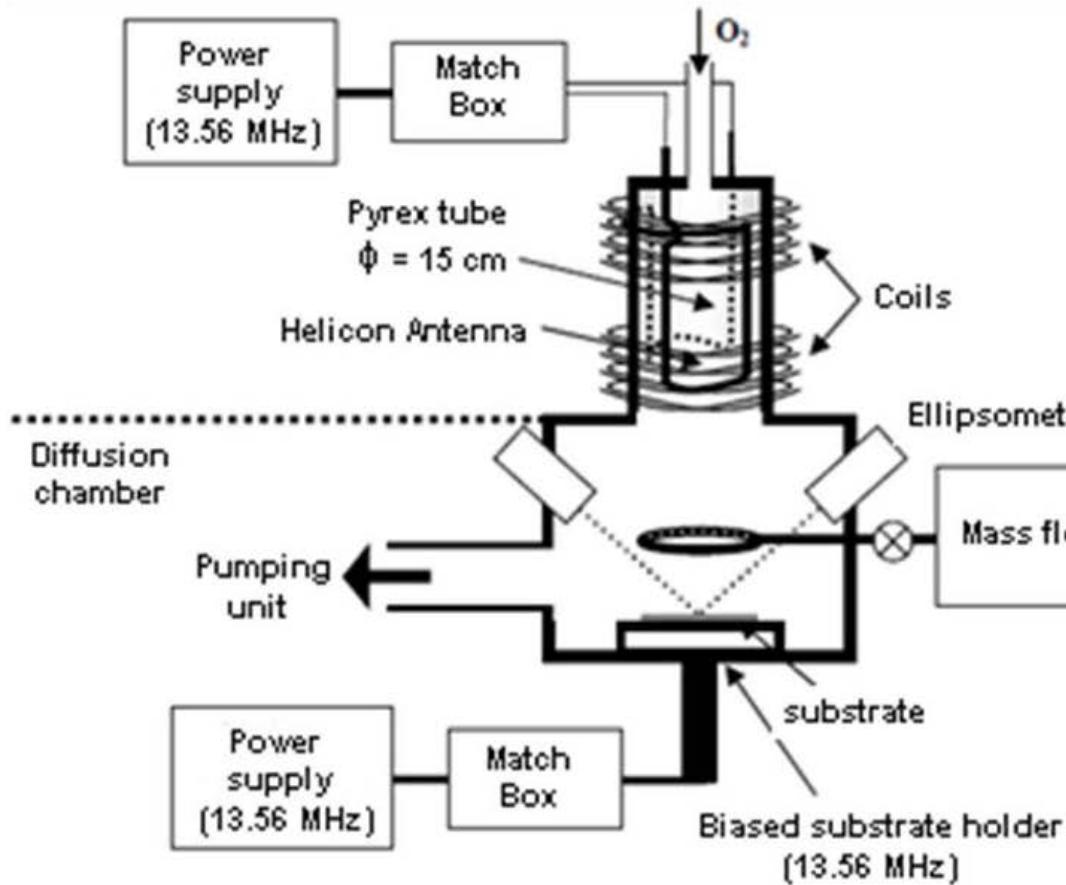
- **Absorption spectroscopy** → measurement of [X] and [X₂] (G. Cunge et al for Cl and Cl₂)
- **Fluorescence spectroscopy** → Radial profile of [X] (JP Booth et al, in CF₄ based plasmas)
- **Spinning wall method** (L. Stafford and V. Donnelly, in Cl₂ plasmas)
- **Time resolved optical emission spectroscopy in pulsed plasma** (A. Bouchoule and P. Ranson, A. Rousseau et al in H₂ plasmas, A. Granier et al in O₂ and O₂/organometallic plasmas)

V.M. Donnelly, J. Guha, L. Stafford, JVSTA. 010801(2011)

Exemple of determination of γ in a low pressure ICP plasma by time resolved emission spectroscopy (TROES) in pulsed O₂, plasmas

Low pressure RF ICP PECVD reactor

O₂/Ar 95:5



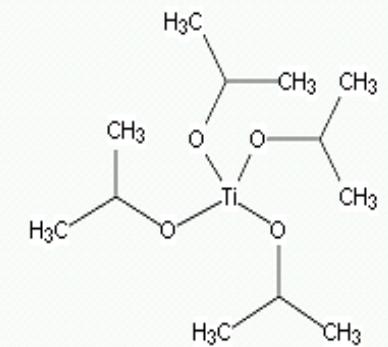
p : 3 mTorr

P_{RF source} : 400 W

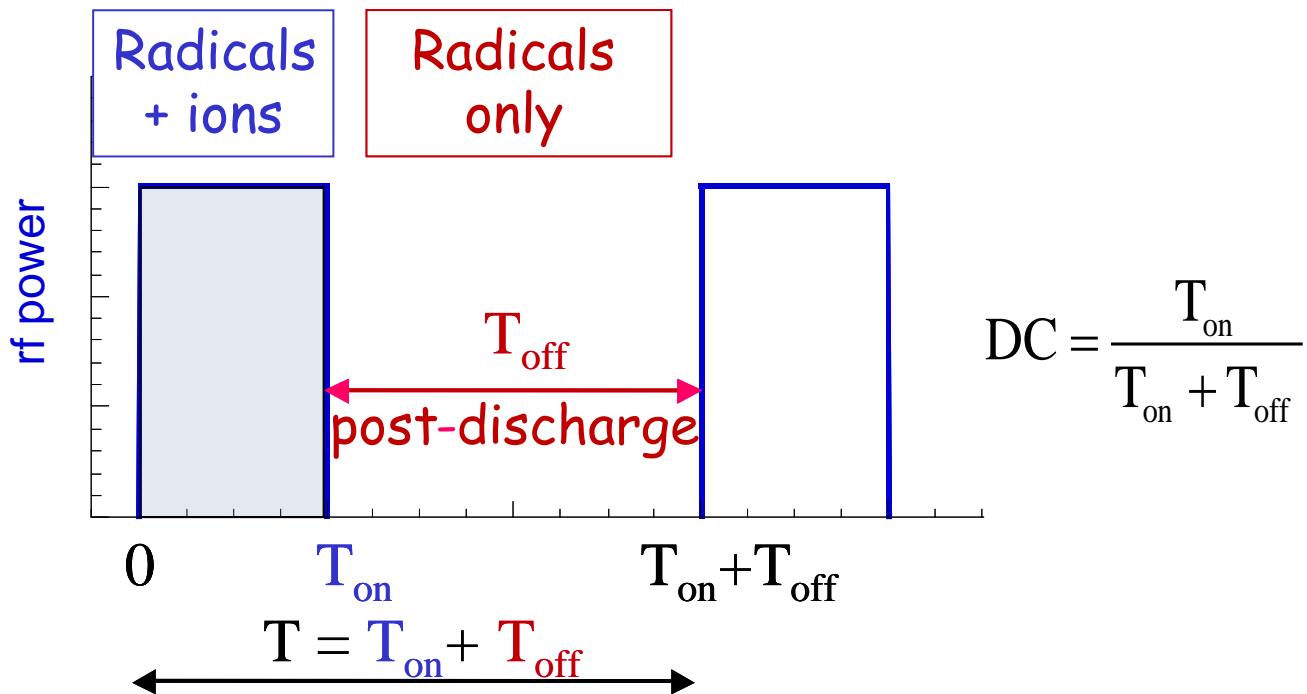
CW and pulsed mode

O₂/HMDSO/TTIP plasmas

TTIP



Characteristic times in pulsed plasmas



- Characteristic times in low pressure plasmas ($1 \text{ mTorr} < p < 1 \text{ Torr}$):
- Electron temperature rise and decay : $10\text{-}50 \mu\text{s}$
- Radical lifetime : $1\text{-}10 \text{ ms}$
- Residence time of stable species : $100\text{-}1000 \text{ ms}$

1 Hz - 100 kHz

O atom kinetics (O in fundamental state)



$O + \text{paroi} \rightarrow \frac{1}{2} O_2$ $K_{\text{paroi}} = \text{loss frequency of } O \text{ at the wall},$
 $K_{\text{paroi}} = 1/\tau, \quad \tau : \text{characteristic loss time}$

Plasma created in continuous mode (CW) :

$$[O] = 2 K_{\text{diss}} \times n_e \times [O_2] / K_{\text{paroi}} = t \times K_{\text{diss}} \times n_e \times [O_2]$$

Plasma created in pulsed mode :

$$[X]_{\text{on}}(t) = [X]_{\text{max}} \cdot \left[1 - \exp\left(-\frac{t}{\tau_{\text{on}}}\right) \right]$$

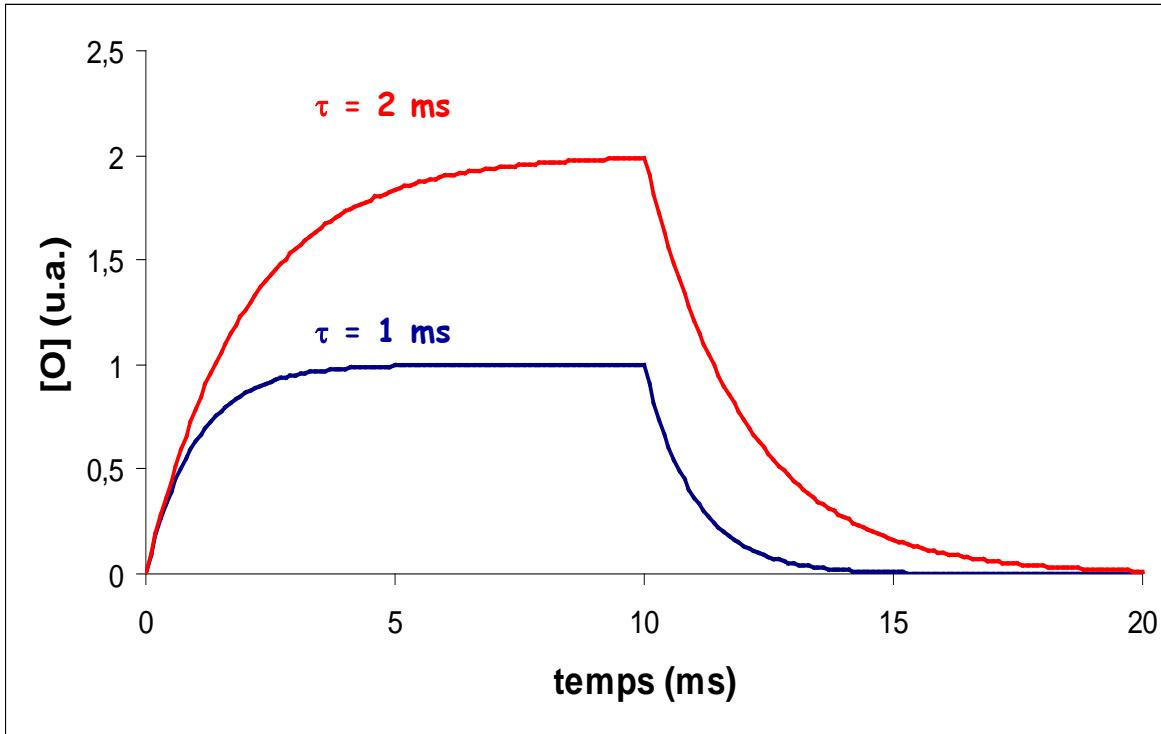
$$[X]_{\text{off}}(t) = [X]_{\text{max}} \cdot \exp\left(-\frac{t}{\tau_{\text{off}}}\right)$$

Kinetics of X atoms in fundamental state

if τ is doubled (i.e. K_{paroi} divided by 2)

→ creation of X 2 times more slowly

→ density of X at the stationnary state multiplied by 2



Determination of [O] from I_O/I_{Ar}

O_2 (+ 5% Ar) - O line at 844.6 nm and Ar line at 750.4 nm

hypotheses :

- maxwellian EEDF, T_e measured in O_2 plasma ($T_e = 3.5$ eV)
- $O + e \rightarrow O^* + e$ et $O_2 + e \rightarrow O^* + O + e$ (Walkup et al, 1986)
- $Ar + e \rightarrow Ar^* + e$ (Lawrence, 1976)
- O^* and Ar^* lost by radiative deexcitation
- $C(\lambda)$ response of the optical system (measured)

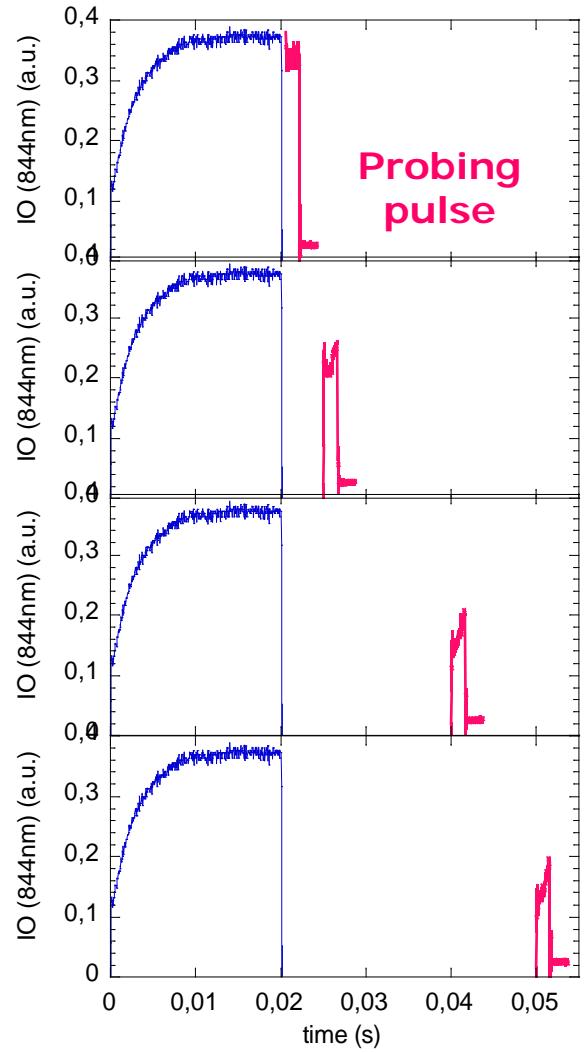


$$\frac{I_O}{I_{Ar}} = \frac{C(\lambda_O)}{C(\lambda_{Ar})} \times \frac{\nu_O}{\nu_{Ar}} \times \frac{k_{exc}^O [O] + k_{diss}^{O_2} [O_2]}{k_{exc}^{Ar} [Ar]}$$

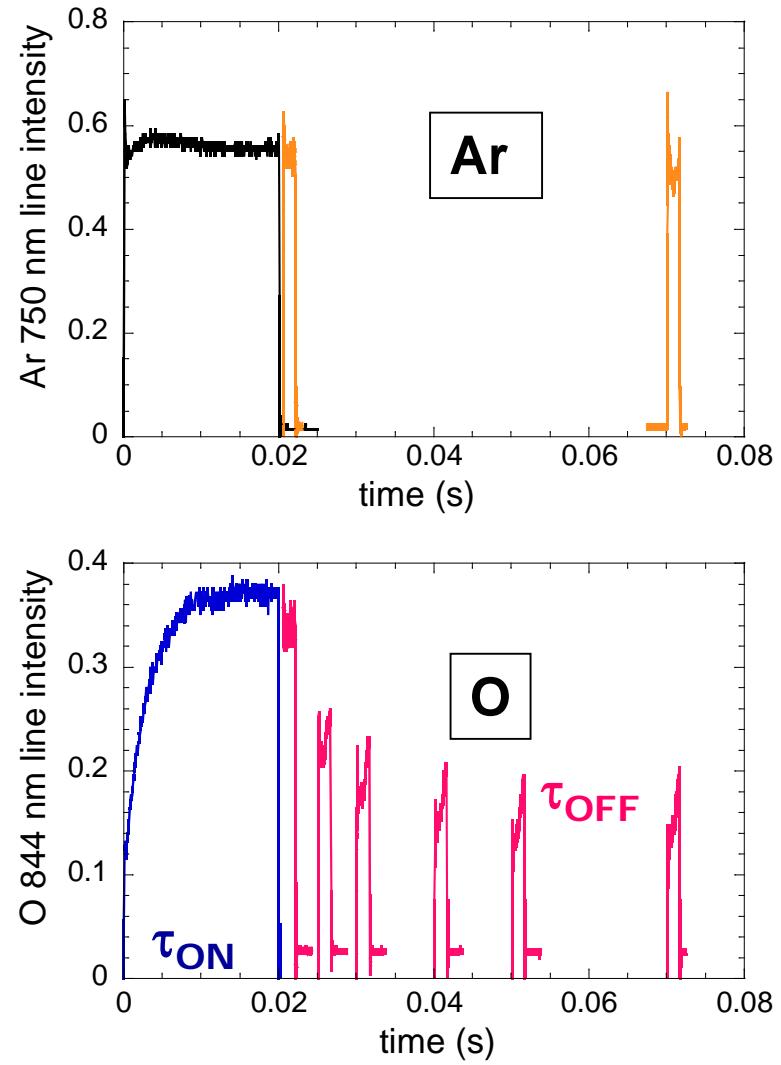
A. Granier et al , J. Appl. Phys 75, 101 (1994)

A. Granier et al, Plasma Sources Sci. Technol. 97, 147 (1997)

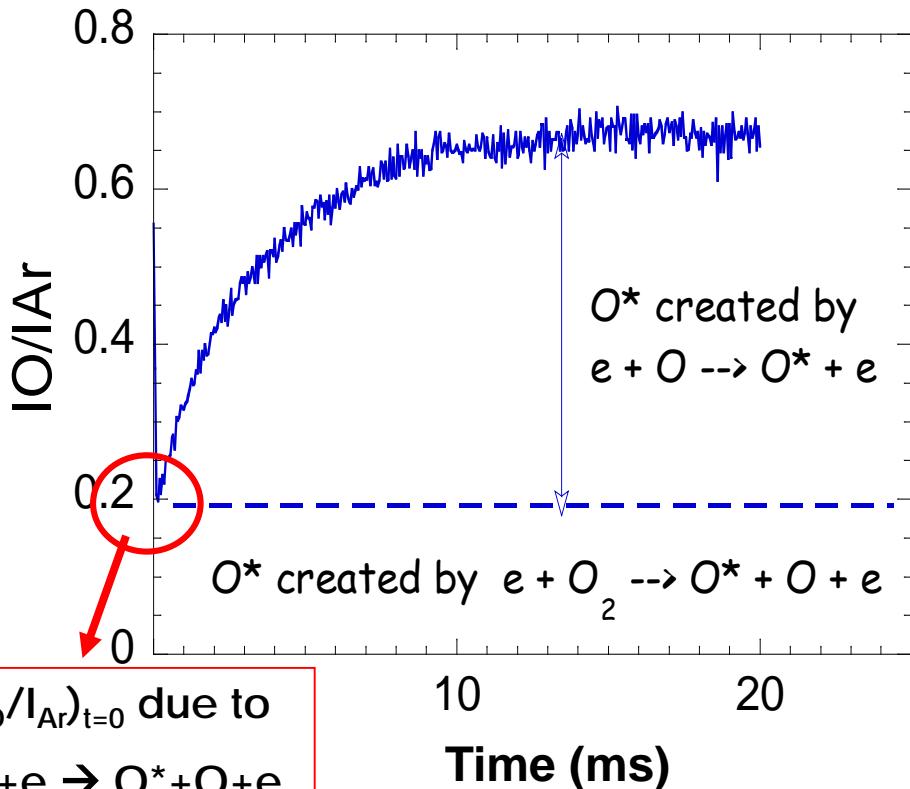
Time Resolved Optical Emission Spectroscopy



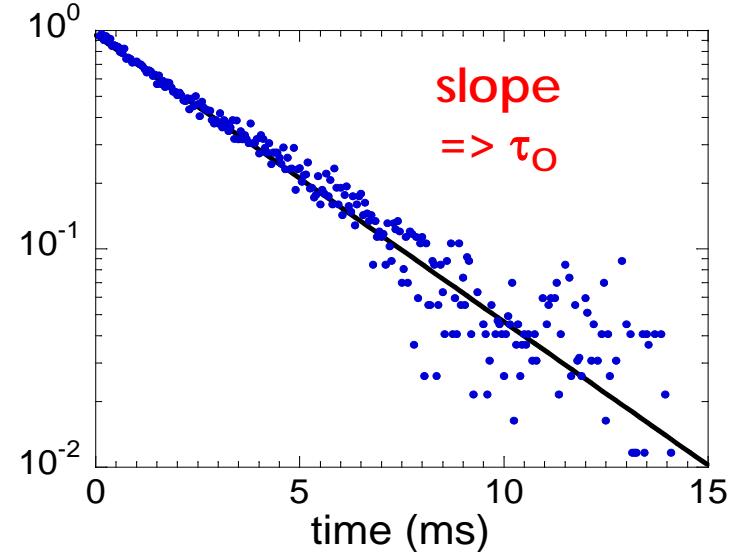
A. Bouchoule, P. Ranson - J. Vac. Sci. Technol. A 9(2) (1991) 317.



Determination of γ_O from $I_O \text{ 844nm} / I_{\text{Ar } 750 \text{ nm}}$ (T_{on})



$$\frac{I_O}{I_{\text{Ar}}} = \frac{C(\lambda_O)}{C(\lambda_{\text{Ar}})} \times \frac{\nu_O}{\nu_{\text{Ar}}} \times \frac{k_{\text{exc}}^O[O] + k_{\text{diss}}^{O_2}[O_2]}{k_{\text{exc}}^{\text{Ar}}[\text{Ar}]}$$



$$\tau_{\text{on}} = \tau_{\text{diff}} + \tau_{\text{loss}}$$

$$\tau_{\text{on}} = \frac{\Lambda_0^2}{D} + \frac{V}{A} \cdot \frac{2(2 - \gamma_{\text{on}})}{\bar{v} \cdot \gamma_{\text{on}}}$$

Chantry et al J. Appl. Phys. 62, 1141 (1987)

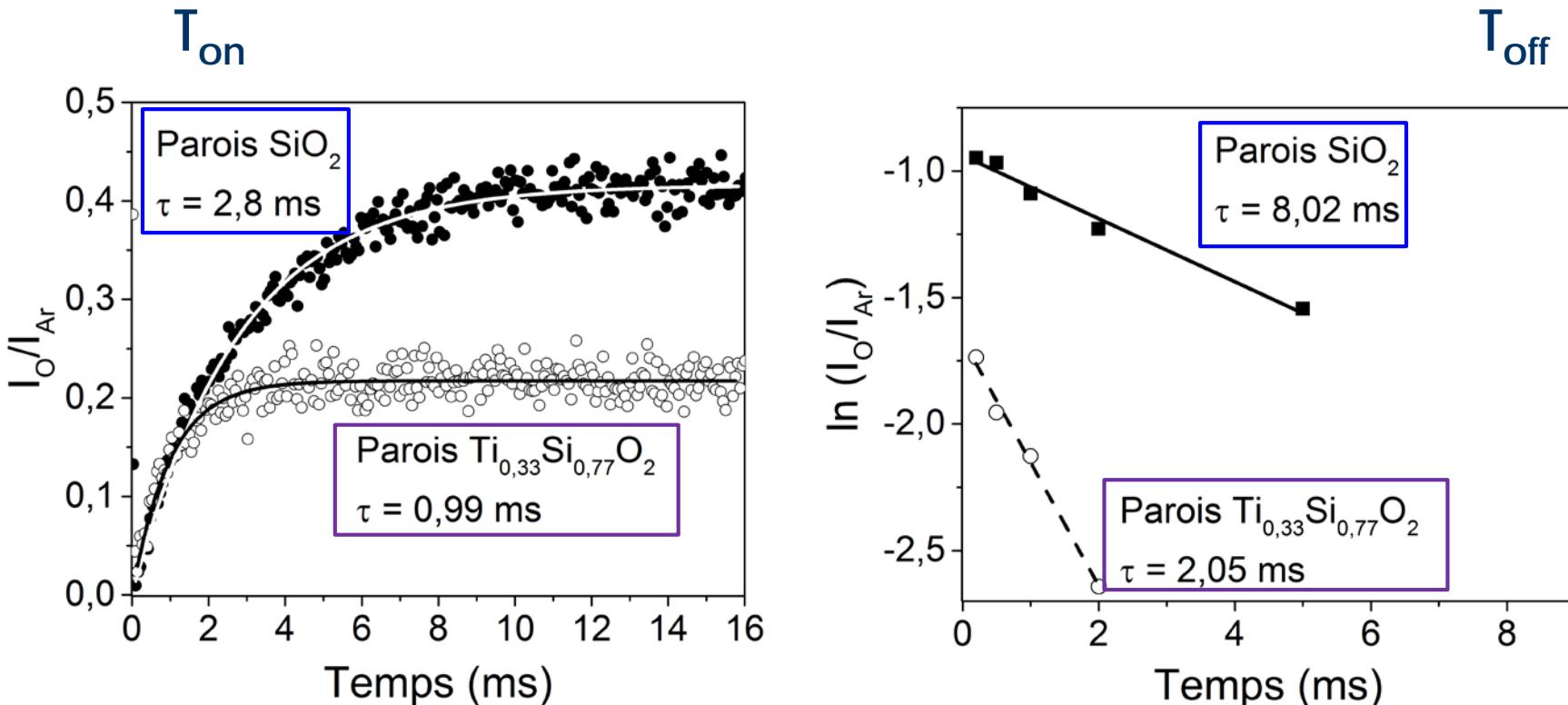
→ γ_O

Oxygen plasma created under the same conditions (pressure, power) with the walls covered with different materials:

- SiO_2 (deposited in O_2/HMDSO plasmas)
- TiO_2 (deposited in O_2/TTIP plasmas)
- $\text{Ti}_x\text{Si}_{1-x}\text{O}_2$ (deposited) in $\text{O}_2/\text{HMDSO}/\text{TTIP}$ plasmas

Influence of the wall nature on O kinetics in O₂ plasmas (T_{on})

5 Hz, T_{on} = 20 ms, T_{off} = 180 ms



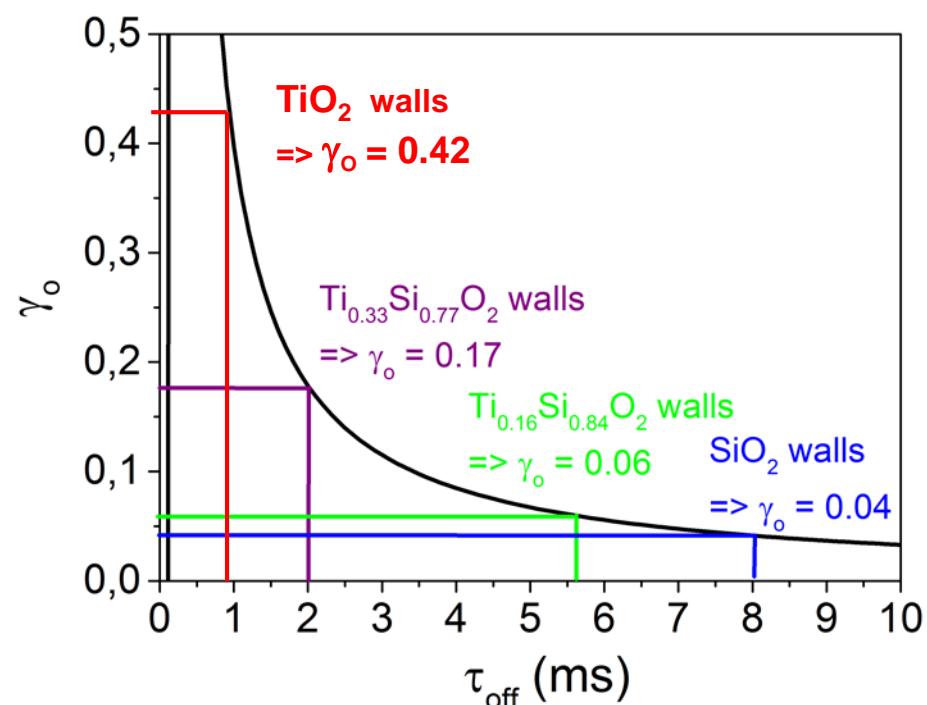
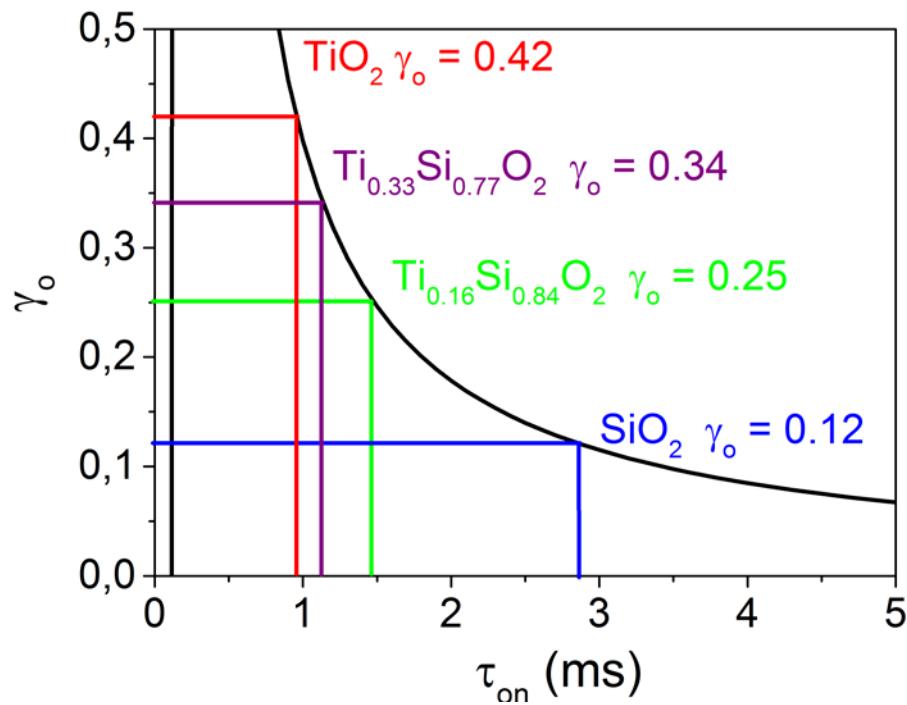
Thèse A. Bousquet (2005), Thèse de S. Elisabeth (2015)

A. Bousquet, A. Goulet, A. Granier dans "Plasmas Froids. Interactions Plasma-Surface : Modèles, Diagnostics et Procédés". Collection : Intégrations, Vol 7 – Ed. G Lelièvre, MRCT CNRS 2011 ISBN 978-2-918701-05-7 (2011), pp 211-234.

Influence of the wall nature on γ_o in O₂ plasmas (T_{on} & T_{off})

Chantry et al J. Appl. Phys. 62, 1141 (1987)

$$\tau = \tau_{diff} + \tau_{loss} = \frac{\Lambda_0^2}{D} + \frac{V}{A} \cdot \frac{2(2-\gamma)}{\bar{v}\cdot\gamma}$$



Thèse de S. Elisabeth (2015)

A. Bousquet, G. Cartry, A. Granier, Plasma Sources Science and Technology 16, 597-605 (2007)

Influence of the wall nature on γ_O in O₂ plasmas (T_{off})

wall material	γ_O (T _{on})	γ_O (T _{off})
TiO ₂	0.42 ± 0.2	0.42 ± 0.2
Ti _{0.33} Si _{0.67} O ₂	0.34 ± 0.1	0.17 ± 0.05
Ti _{0.16} Si _{0.84} O ₂	0.25 ± 0.1	-
SiO ₂	0.12 ± 0.02	0.04 ± 0.01

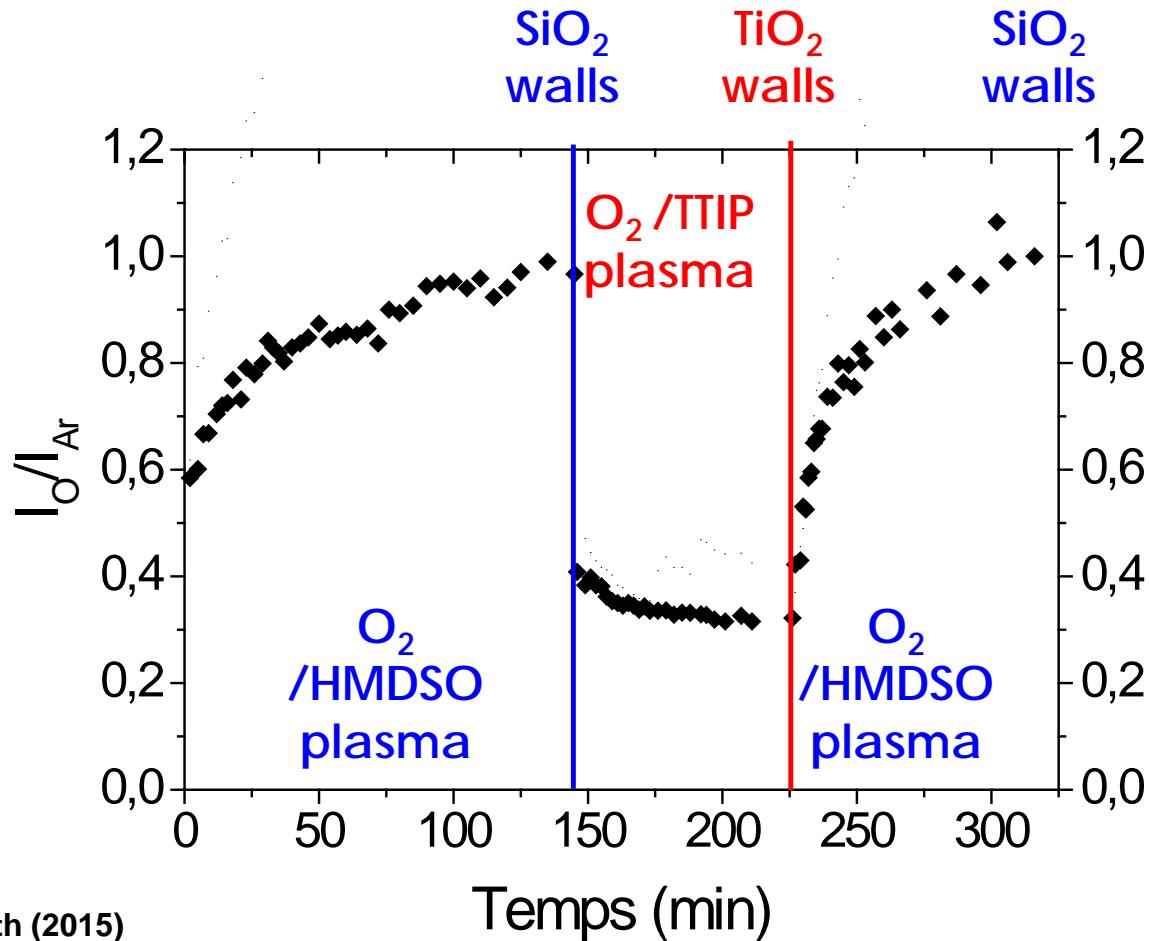
- γ_O (T_{on}) decreases as Si added to TiO₂, from 0.42 on TiO₂ to 0.12 on SiO₂
- TiO₂ : same (high) value of γ_O during T_{on} and T_{off}
- SiO₂ : γ_O reduced by a factor 3 in the post-discharge
 - consistent with the literature and attributed to ion induced recombination and absence of ion bombardment during the post-discharge

Thèse A. Bousquet (2005), Thèse de S. Elisabeth (2015)

O atom kinetics studied by TROES in pulsed O₂/TIPT and O₂/HMDSO plasmas

In situ demonstration of the strong effect of the nature of the oxide covering the wall by OES : O atoms

I_O/I_{Ar} monitored during successive cycles of $O_2/HMDSO$ and $O_2/TTIP$ plasmas



Thèse de S. Elisabeth (2015)

O atom kinetics in O₂/TTIP and O₂/HMDSO plasmas

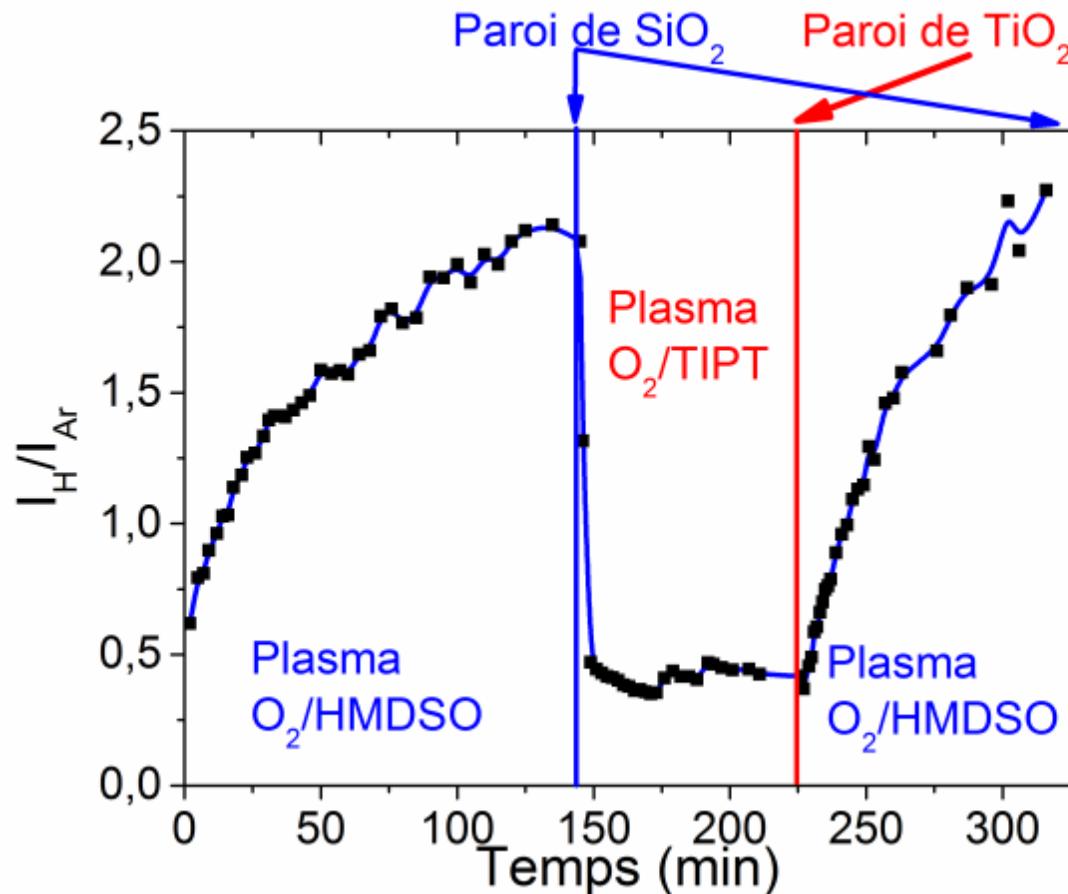
PECVD plasma	γ_O (T_{on})	O ₂ plasma on	γ_O (T_{on})
O ₂ /TTIP 98.5 : 1.5	0.42 ± 0.2	TiO ₂	0.42 ± 0.2
O ₂ /HMDSO 99.5 : 0.5	0.065 ± 0.01	SiO ₂	0.12 ± 0.02

- TiO₂ walls : same value of γ_O in O₂/TTIP plasmas and O₂ plasma
- SiO₂ walls : γ_O in O₂/TTIP plasmas = 0.5 x γ_O in O₂ plasma
 - attributed to OH reactivity on reactive sites which limits O recombination in O₂/HMDSO plasma (consistent with results obtained in O₂/HMDSO 85:15 plasmas, where the effect was stronger due to higher OH contents)

Does it mean that O is more reactive than OH on TiO₂ surfaces while OH more reactive than O on SiO₂ surfaces ?

In situ demonstration of the strong effect of the nature of the oxide covering the wall by OES : H atoms

I_H/I_{Ar} monitored during successive cycles of $O_2/HMDSO$ and $O_2/TIPT$ plasmas



Thèse de S. Elisabeth (2015)

H atom kinetics in O₂/TTIP and O₂/HMDSO plasmas

PECVD plasma	γ_H (T_{on})
O ₂ /TTIP 98.5 : 1.5	0.065 ± 0.2
O ₂ /HMDSO 99.5 : 0.5	0.01 ± 0.002

γ_H in O₂/TTIP plasmas (TiO₂ walls) = 6 x γ_H in O₂ /HMDSO plasma (SiO₂ walls)

Conclusion

Real difficulty to predict values of recombination frequencies

Atom densities can vary on very large scales according to the nature of the walls for the same discharge parameters (power, pressure...), typically from 1% to several ten percent

Pulsed plasma : for the same « walls » γ values most often different in discharge and post-discharge times